

$$W = \frac{m \text{ L-Se (bakter.)}}{m \text{ neu p-p-e}} \cdot 100\%$$

$$m_{\text{ndf.ppe}} = m_{\text{ppe1}} + m_{\text{ppe2}} - m_{\text{ndf.refe.}}$$

$$m_{\text{res. ppe}} = m(\text{ac. CO}_2) + m_p(\text{H}_2\text{O}) - m(\text{CO}_2)$$

$$J(\text{BaCO}_3) = \frac{1.972}{1.972} = 0.01 \text{ mmol}$$

$$m_{\text{pre}}(\text{HBr}) = 150 \text{ gms} \cdot 1,180 \text{ gms/l} = 168 \text{ g}$$

$$m_{\text{L-L}}(\text{HBr}) = \frac{168 \cdot 20\%}{100\%} = 33,62$$

9/11/85	2	33,62	2	9,45 евро
		85/1000		

$$\alpha = 0,415 : 2 = 0,2075 - \text{наго ВЭОБ, где } I \text{ неперекрывает}$$

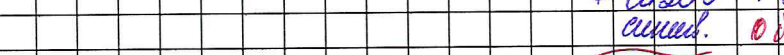
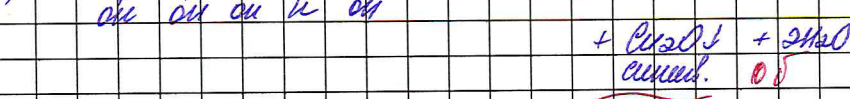
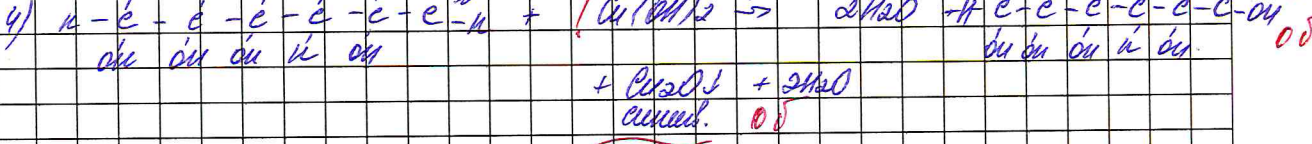
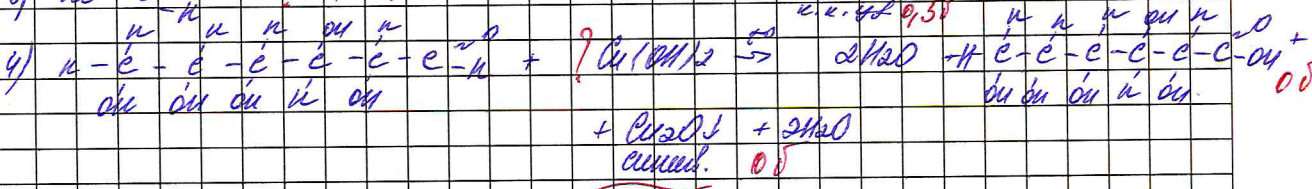
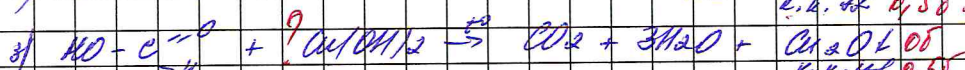
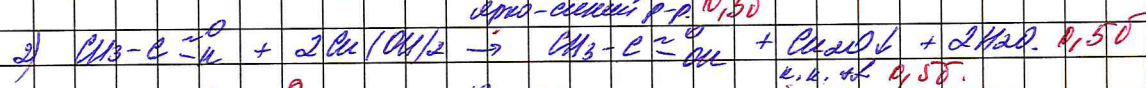
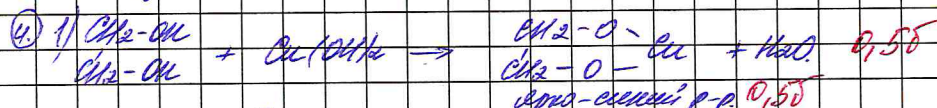
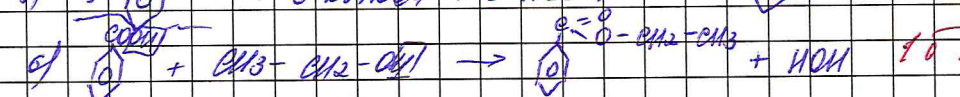
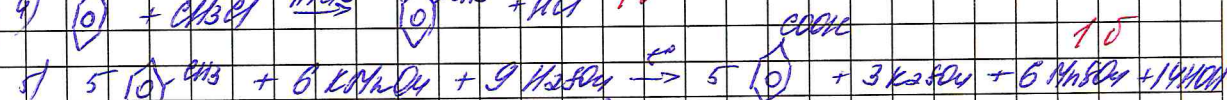
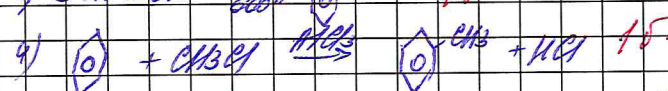
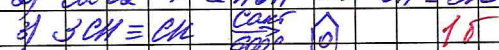
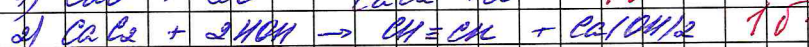
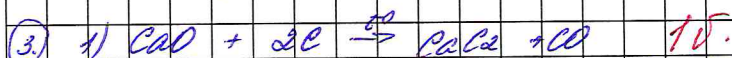
$$m(\text{CO}_2) = 0,01 \text{ моль} \cdot (12 + 16 \cdot 2) = 0,442$$

$$m(\text{ball}) = P.M = 0,01 \text{ mola} \cdot (137 + 80 \cdot 2) = 2,97 \text{ g}$$

$$\text{m now pre} = 1,972 + 1682 - 0,442 = 169,53$$

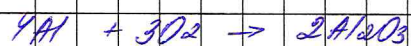
$$\omega = 2,942 \cdot 100\% = 1,752\%$$

Order: 1.452%

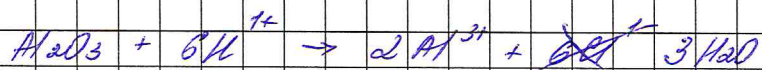
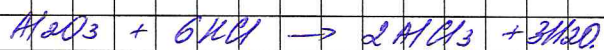


Шифр 11-14

(5.)  $Al_2O_3$  - амфот. оксид 3 периода



05



Уточно: 13,58.

РМБ - Крикуновско Л.Л.  
И - Чанско Л.Л.